

Tetrazole anchored triazole: Potential structure and function directing tecton in Metal-organic frameworks

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Dynamic microporous metal-organic frameworks (MOFs) which offer unlimited opportunity in terms of architecture, porosity, endurance and properties have found applications in separation science, gas storage technology, sensors, catalysis, drug delivery etc. ^[1]. In continuation of our modular approach to functional metal-organic frameworks,^[2] we introduce two different tectons: the first based on tetrazole, L1 = 5-(4H-1,2,4-triazol-yl)-2H-tetrazole^[3] and the second based on a natural amino acid, L2 = 3-hydroxy-2-[1,2,4]triazole-4-yl-propionic acid, as a bi-functional bridging ligand and as anion. Synthesis involving ternary components of L1, zinc and phen/phen-dione leads to a 2D network [Zn(phen)₂(L1)]BF₄ (1) which crystallizes in a centrosymmetric space group (Pnma). A mononuclear inclusion complex [Zn(2,2'-bpy)₃]₂(L1)(BF₄)₃·10H₂O (**2**) that crystallizes in a non-centrosymmetric space group (P2₁2₁2₁) was also obtained.

We have studied gas adsorption/storage applications and fluorescence properties of these materials. TGA-DTA of **1** shows exceptional thermal stability up to 200°C whereas **2** which is dehydrated at 100°C, begins to collapse after 110°C. **1** shows little preference for both $N_2(g)$ and $H_2(g)$. Type-I isotherm with a low specific surface area of 11 m²/g for N₂(g) and a non-reversible low H₂(g) uptake of 8 cm³/g was evidenced. In the inclusion complex **2**, the hanging-up of **L1** in the matrix dictates a larger separation of mononuclear entities thus creating sufficient spaces in the host crystal packing. This structural impact is seen in the BET results. Degassed at 30°C under high vacuum, the activated phase of **2** shows a specific surface area of 40 m²/g from N₂(g) isotherm and a H₂(g) uptake of 33 cm³/g. When subjected to a dynamic vacuum at 110°C the surface area increases to 49 m²/g. Upon excitation ($\lambda ex = 345$ nm), **1** emits strong emissions at 373, 392 nm assigned to intra-ligand fluorescence from the capping ligand whereas the shoulder band at 415 nm arises from a charge-transfer transition.

A comparison between the role of neutral and anionic duality of 1,2,4 amino-acid scaffold and the asymmetric triazole-tetrazole tecton, in these coordination polymers is presented.

Keywords: porous MOFs; amino-acids derivatives; phenazine; 4-R-1,2,4-triazole

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